

# Reliable Protein Folding on Non-Funneled Energy Landscapes: The Free Energy Reaction Path

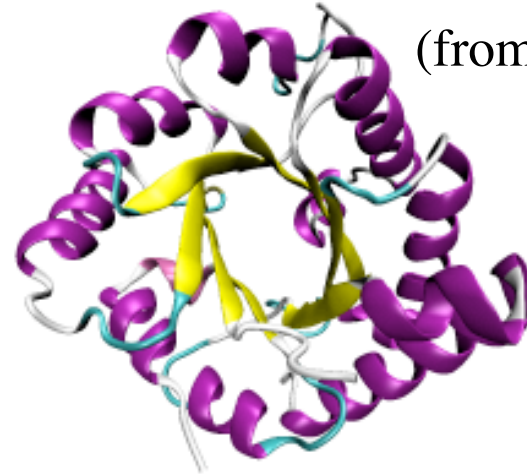
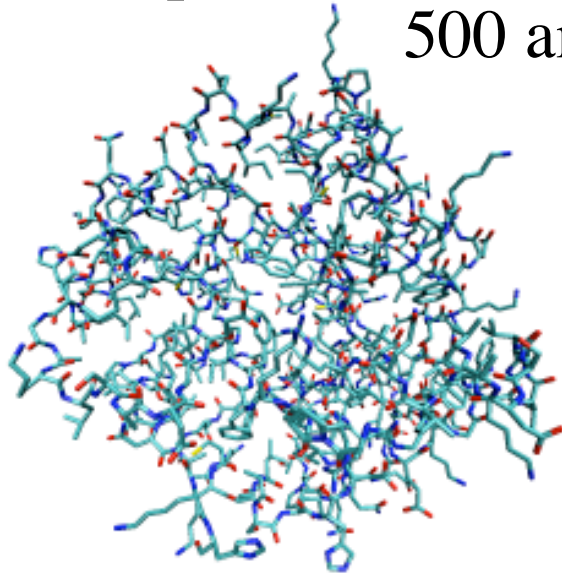
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# Protein Folding

As temperature decreases, a linear chain of amino acids collapses into a reproducible three dimensional structure.

Example:      Triose Phosphate Isomerase  
500 amino acids, folds reliably  
(from Wikipedia)



Levinthal's Paradox (1968): (as paraphrased by Zwanzig, 1992)  
If each amino acid can pair with at least 2 others, and if conformations are randomly sampled no faster than  $10^{15}$  per second.

➔ time to fold  $> 2^{500} \times 10^{-15} \text{ s} \approx 10^{135} \text{ years}$

# Modern View: Funneled Energy Landscapes

Conformation space is not sampled randomly:

In equilibrium,

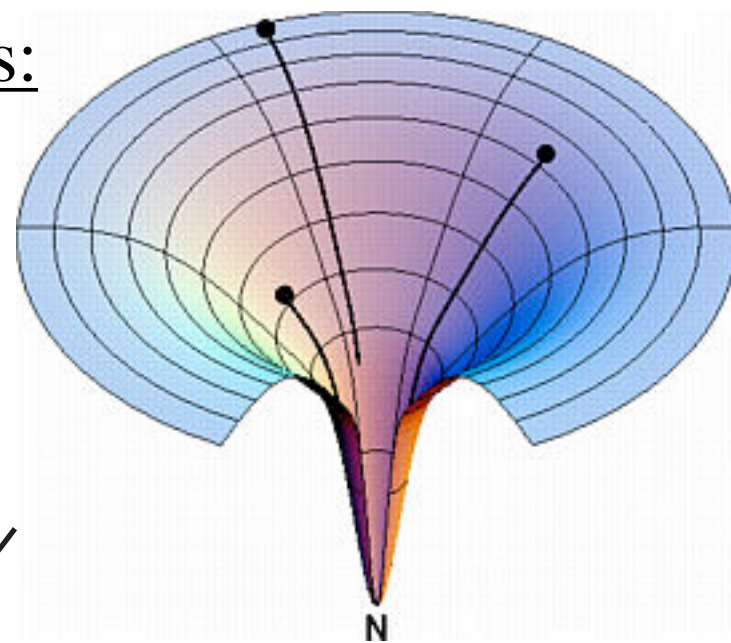
$$\text{Probability} = \exp(-F/T) \quad \text{with} \quad F = E - TS$$

Out of equilibrium, dynamics bias the search.

Energy Landscapes are shaped like funnels:

Out of equilibrium, dynamics go to N

In equilibrium, there is a phase transition

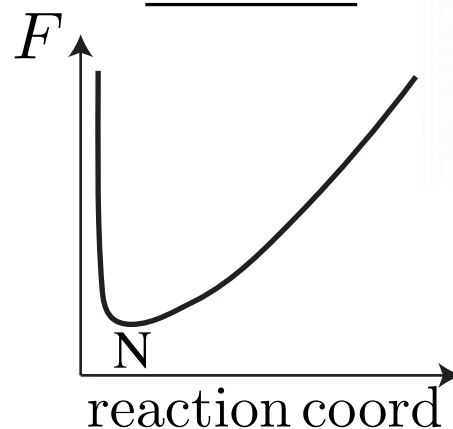
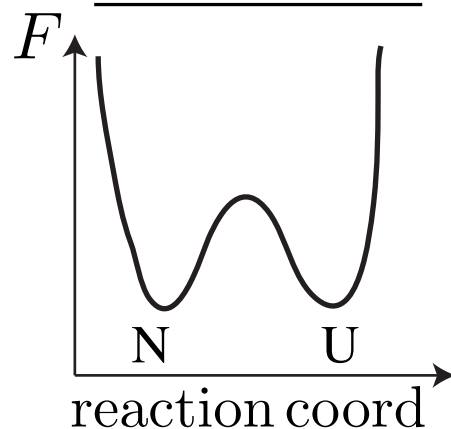
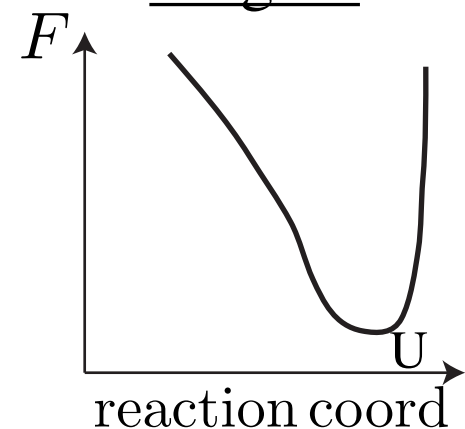


(From Dill & Chan, 1997)

Large T

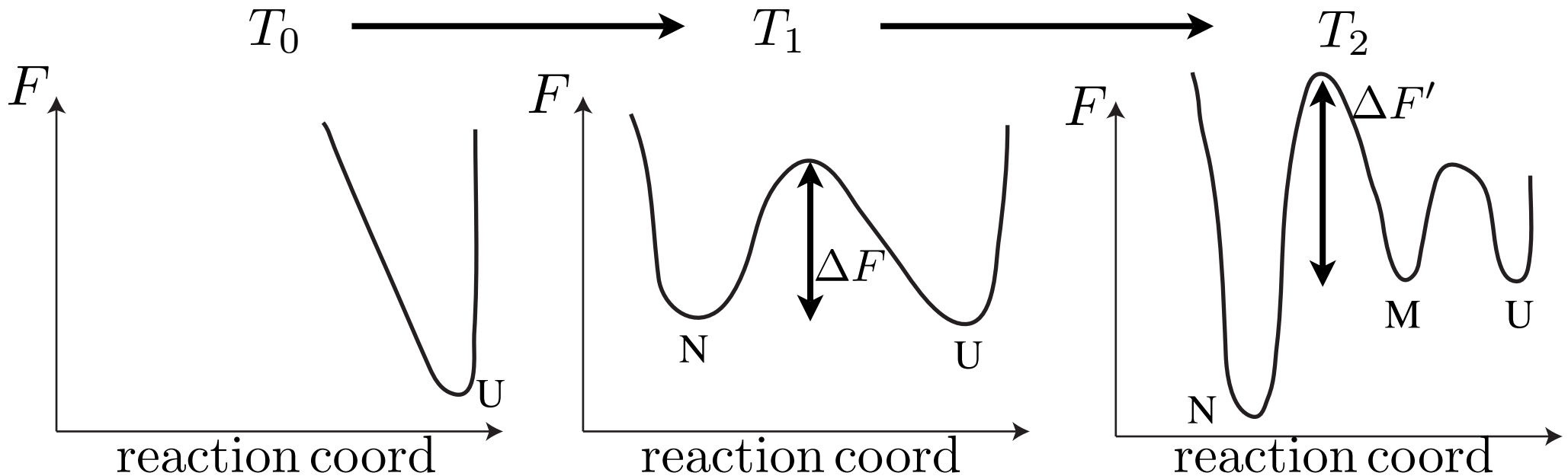
Transition T

Small T



# Folding on Non-Funneled Landscapes

If there are many minima of  $E$  then there are many minima of  $F$



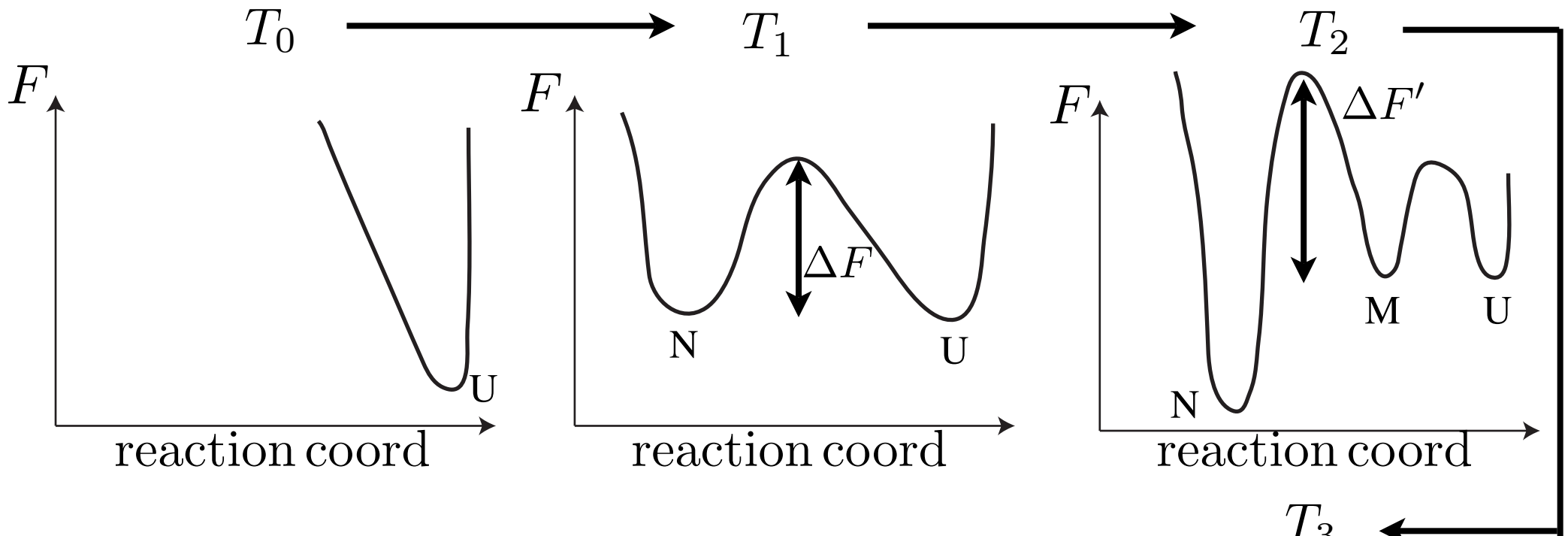
at  $T_1$  the time to fold is  $\propto \exp(\Delta F/T_1)$

Given that  $T(t) = T_0(1 - rt)$  this sets a bound on  $r$ .

If  $r > r^f$  the protein can fall into a metastable state

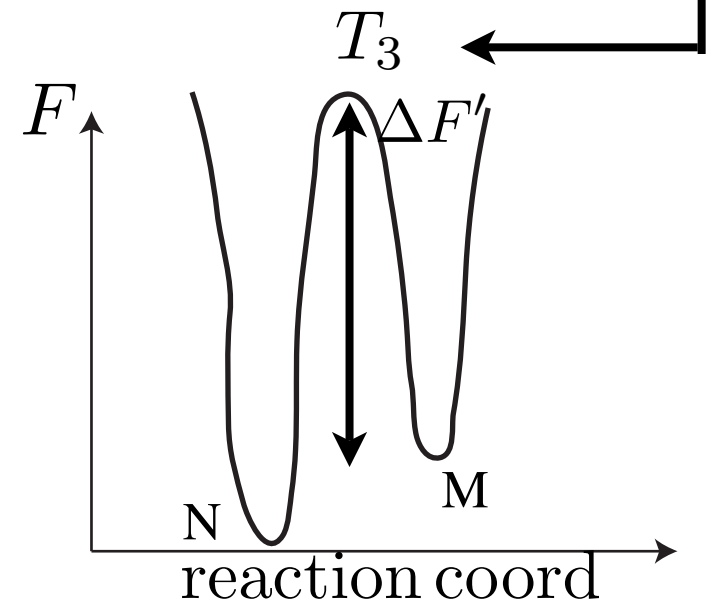
# Folding on Non-Funneled Landscapes

If there are many minima of  $E$  then there are many minima of  $F$



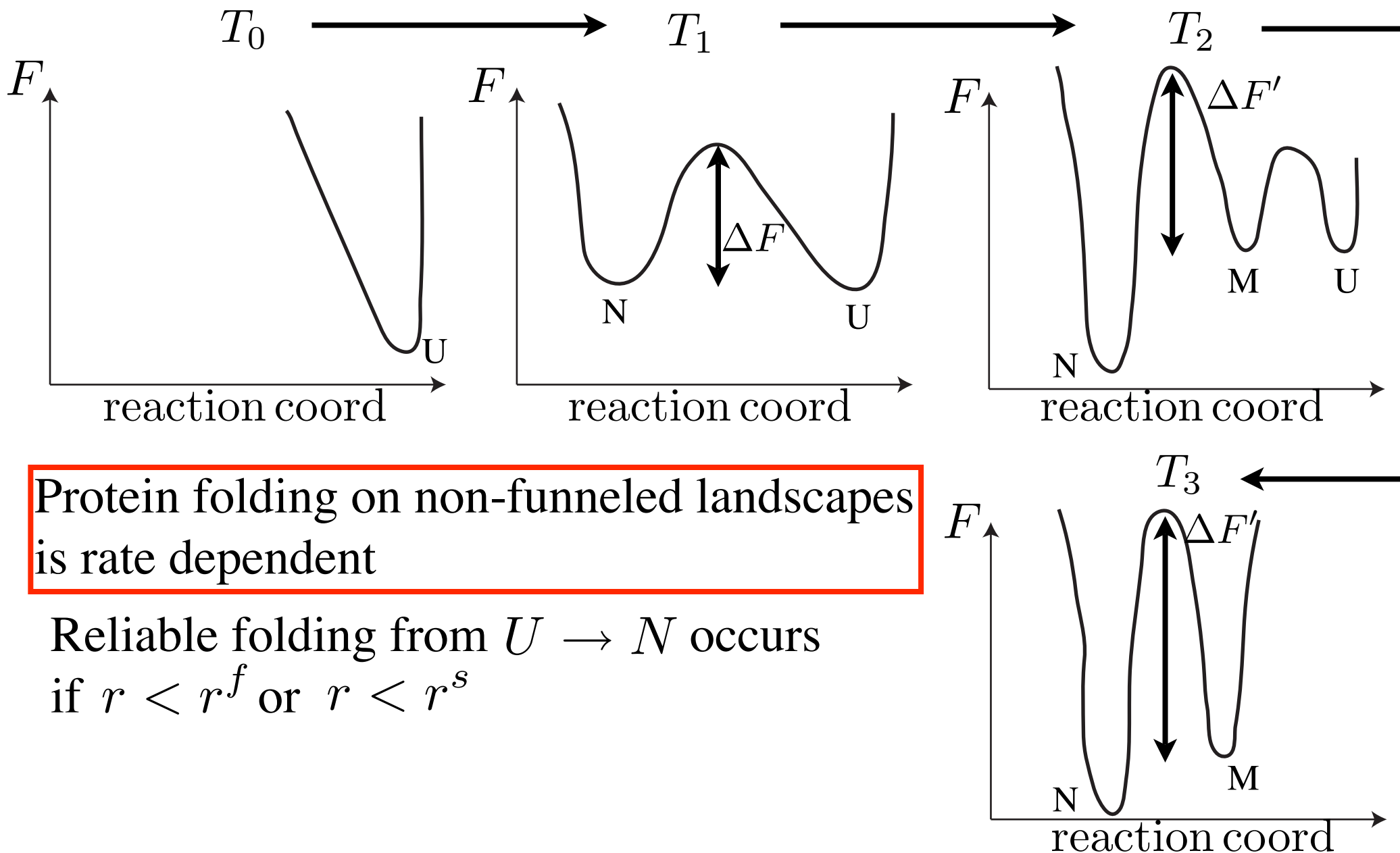
Since  $T(t) = T_0(1 - rt)$ , the barrier  $\Delta F'$  increases faster for large  $r$ .

If  $r > r^s$  the protein will remain in  $M$  indefinitely



# Folding on Non-Funneled Landscapes

If there are many minima of  $E$  then there are many minima of  $F$

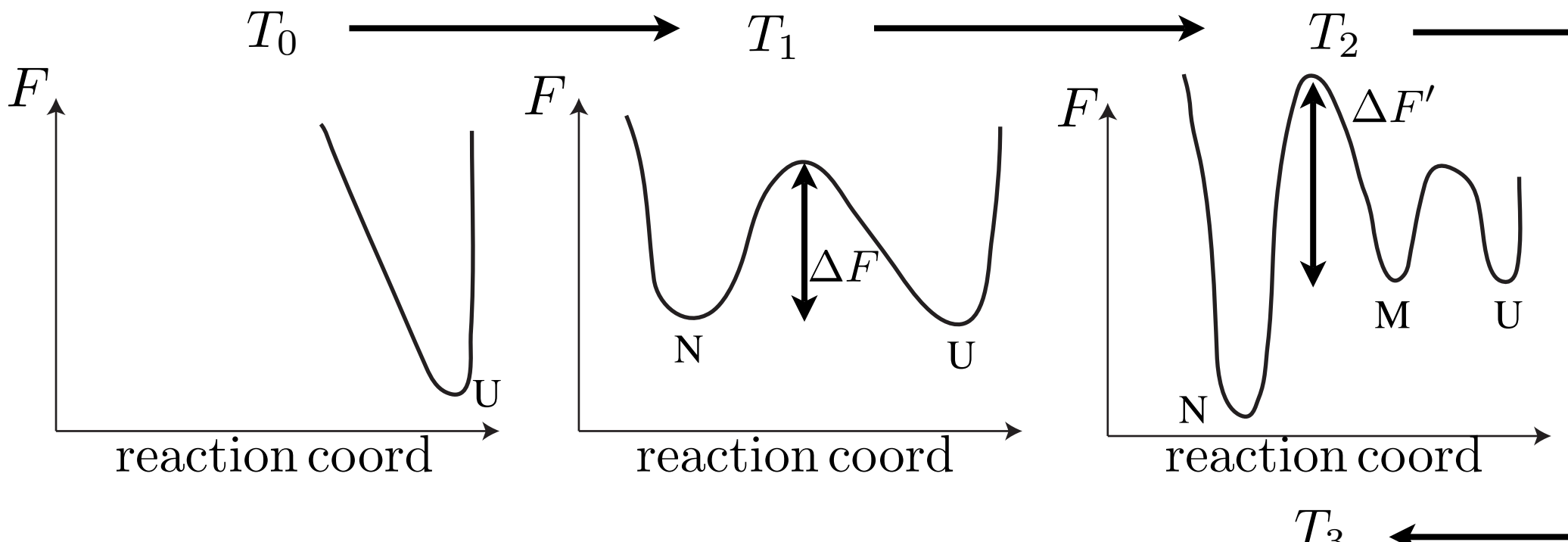


Protein folding on non-funneled landscapes is rate dependent

Reliable folding from  $U \rightarrow N$  occurs if  $r < r^f$  or  $r < r^s$

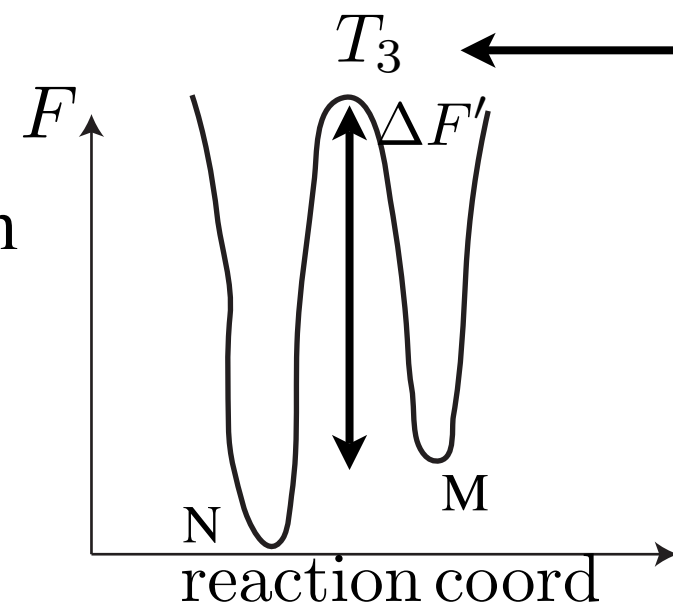
# Folding on Non-Funneled Landscapes

If there are many minima of  $E$  then there are many minima of  $F$



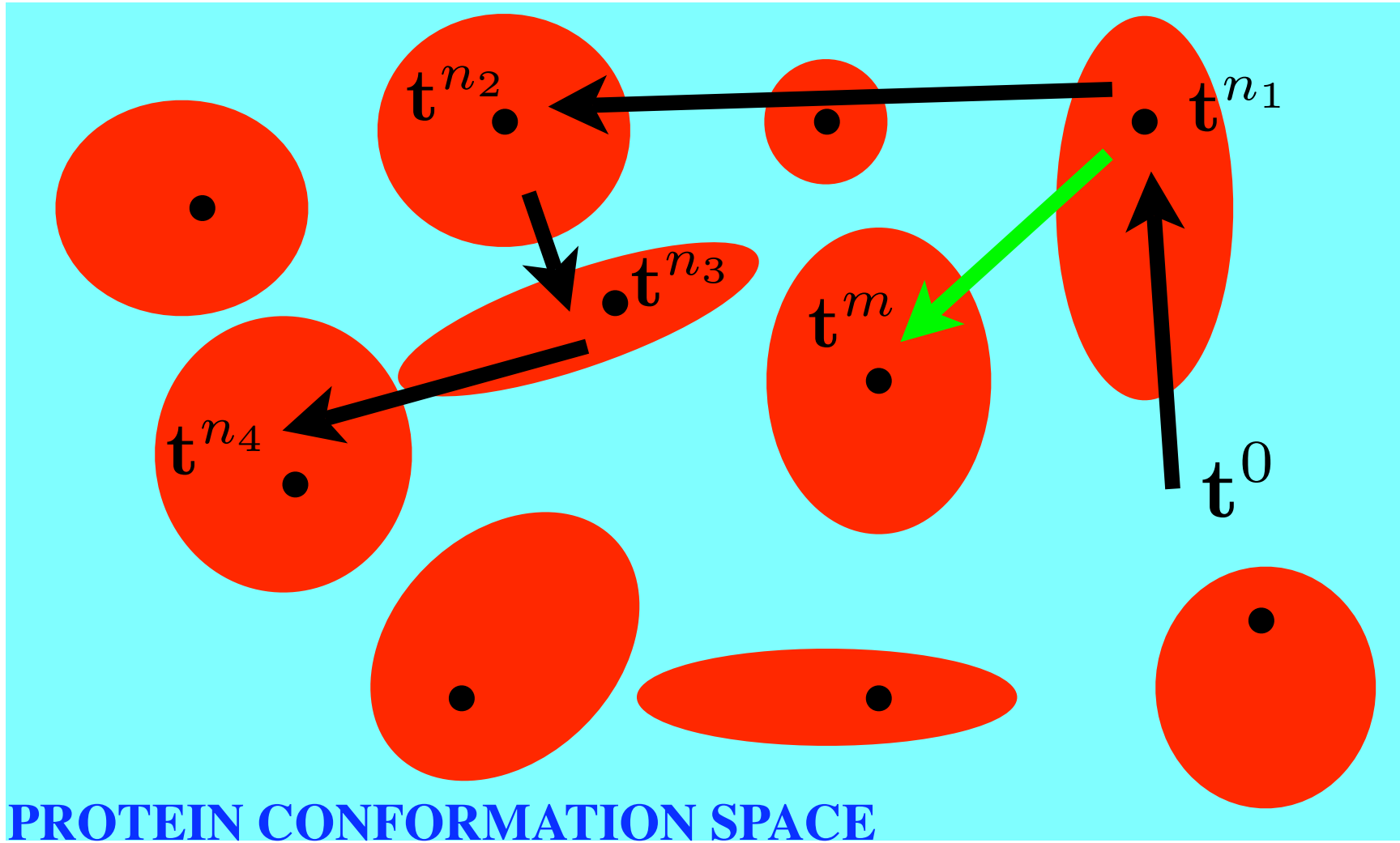
If  $r < r^s$  folding is in equilibrium

If  $r^s < r < r^f$  folding is out of equilibrium



● = topology  $\mathbf{t}^n$   
(energy minima)

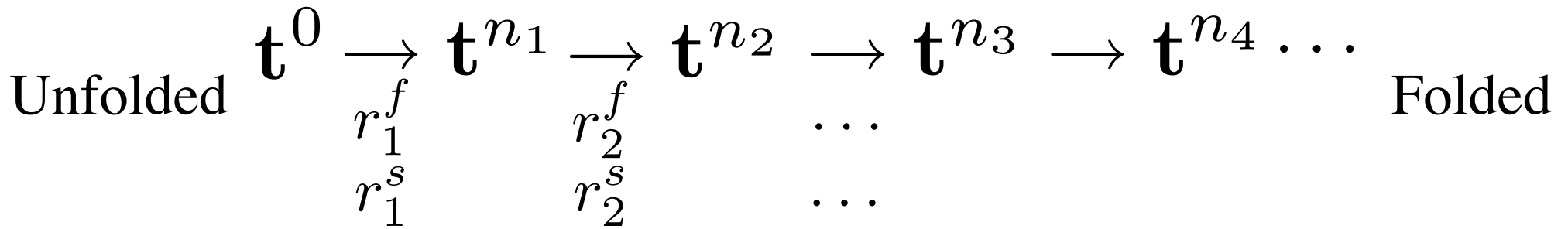
● = basin of attraction  $\mathcal{B}(\mathbf{t}^n)$






In Equilibrium:  $\mathbf{t}^0 \longrightarrow \mathbf{t}^{n_1} \longrightarrow \mathbf{t}^{n_2} \longrightarrow \mathbf{t}^{n_3} \longrightarrow \mathbf{t}^{n_4}$   
 $\mathbf{t}^{n_1} \xrightarrow{\text{green}} \mathbf{t}^m \longrightarrow ?$



# The Free Energy Reaction Path

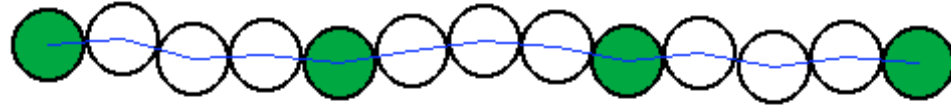


Given  $T = T_0(1 - rt)$

- ① If  $r < r_i^s$  for all transitions  $i$   
 Reliable Equilibrium Folding to global minimum
- ② If  $r_i^s < r < r_i^f$  for a single transition  $i$   
 Reliable Out-of-Equilibrium Folding to  $\mathbf{t}^{n_i}$
- ③ If  $r > r_i^f$  and  $r > r_i^s$  for a single transition  $i$   
 Un-reliable folding

# Simulations

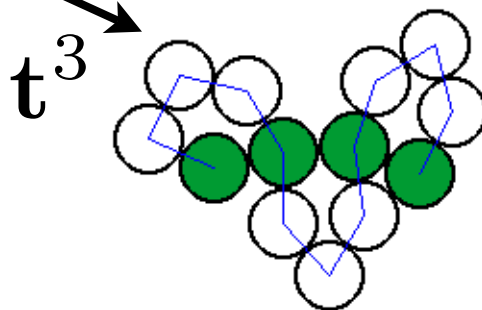
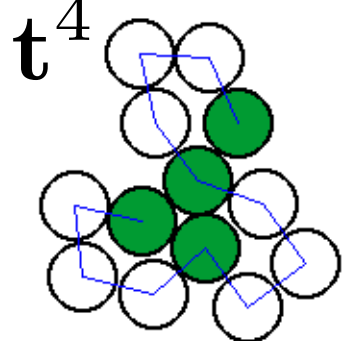
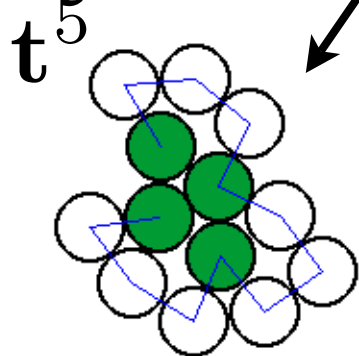
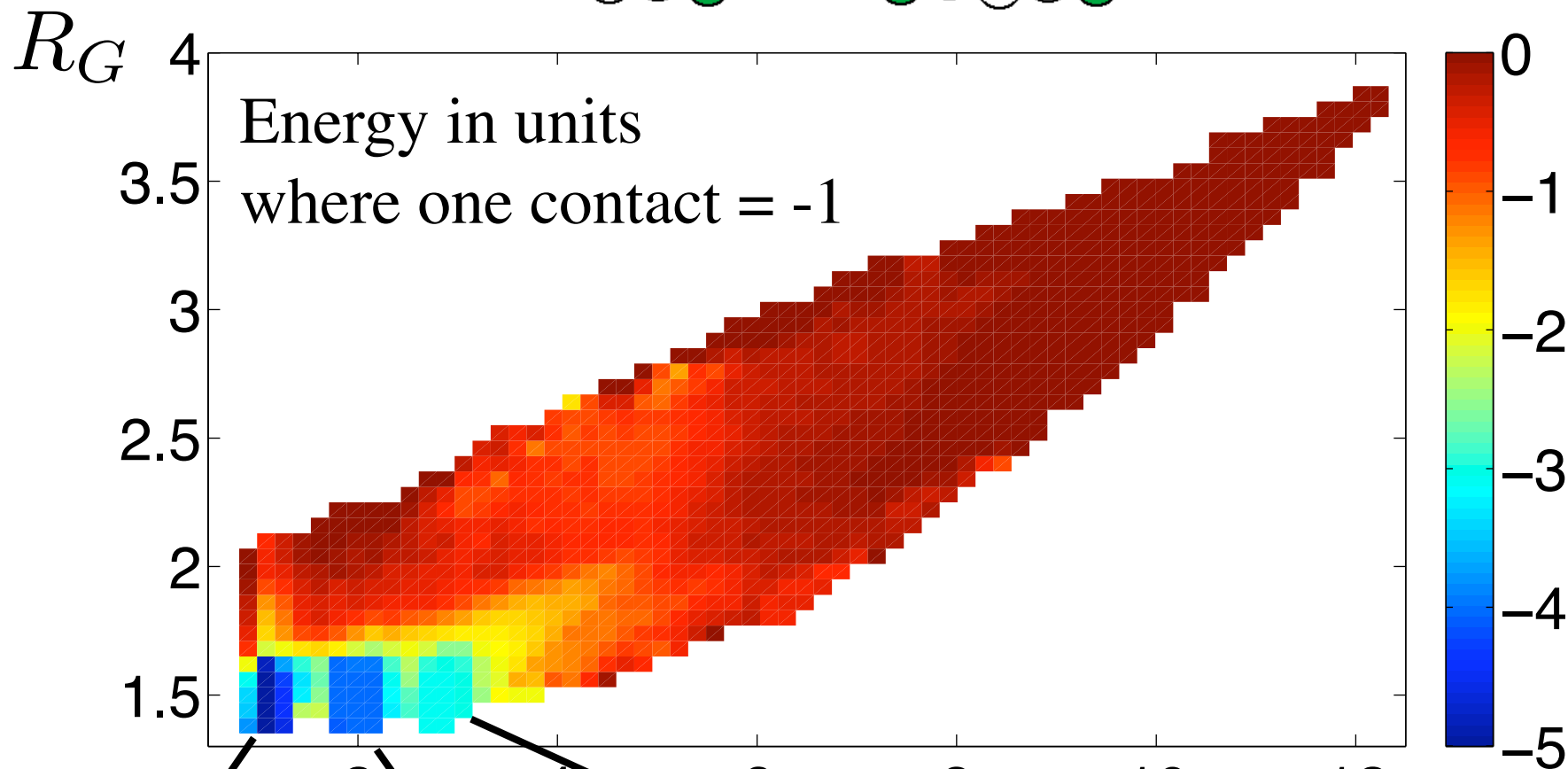
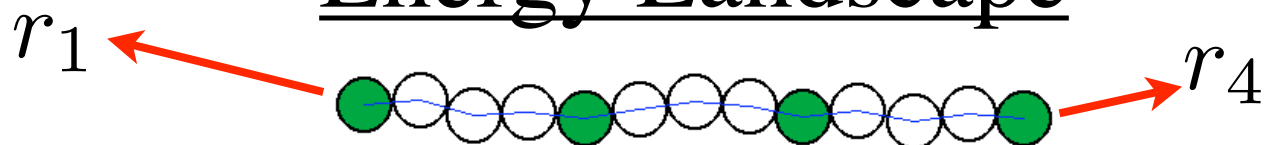
A particular two dimensional  $C^\alpha$  model



Folding is induced by lowering temperature or increasing attractive strength

$$c \equiv \frac{|E_{\text{att}}|}{T} = rt$$

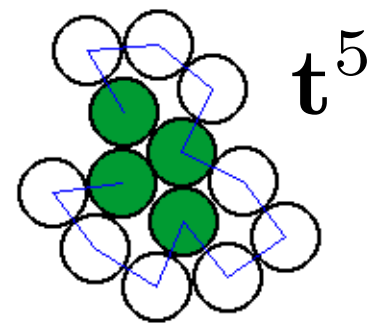
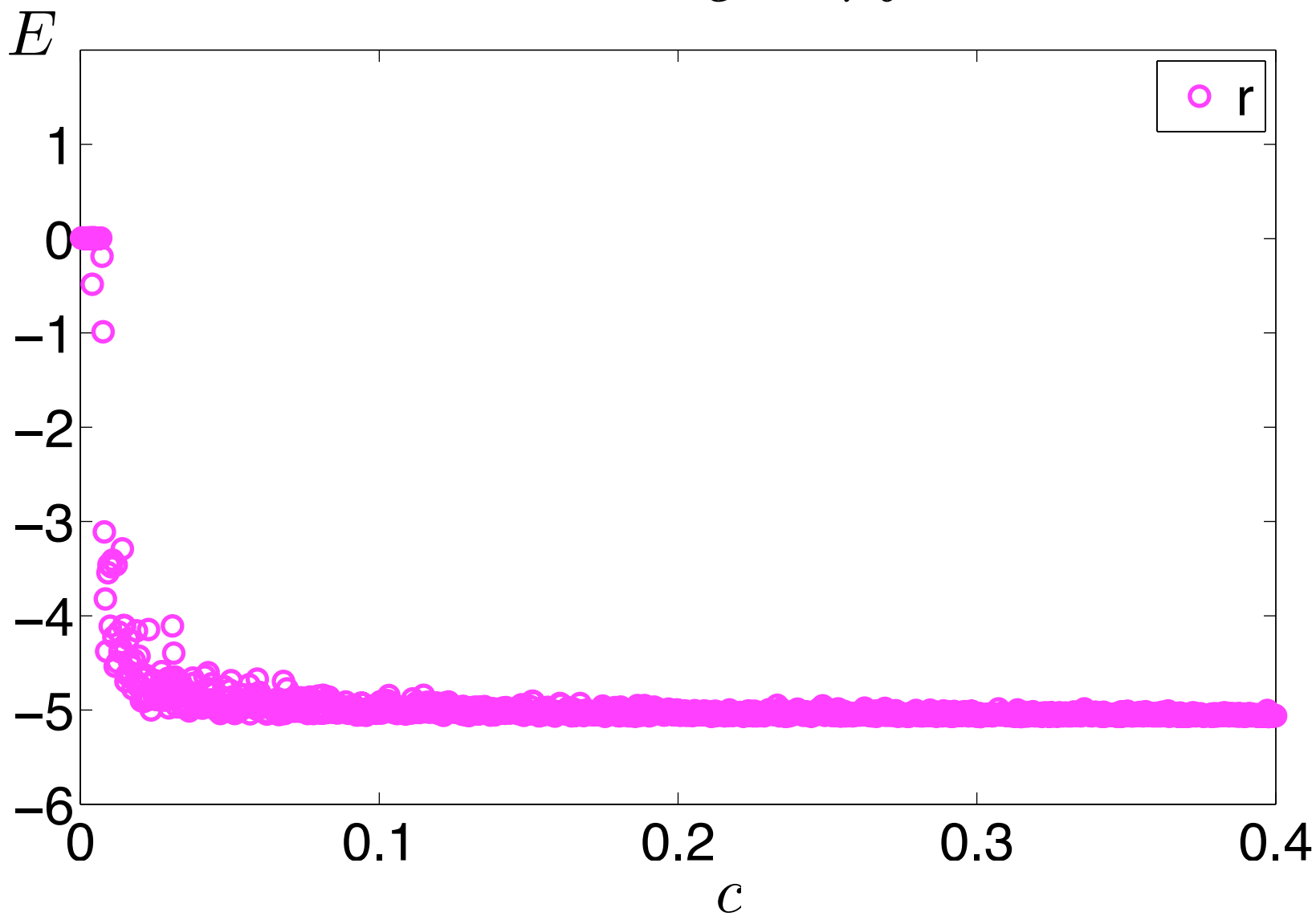
# Energy Landscape



$$|r_4 - r_1|$$

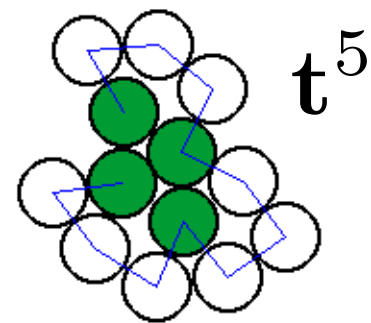
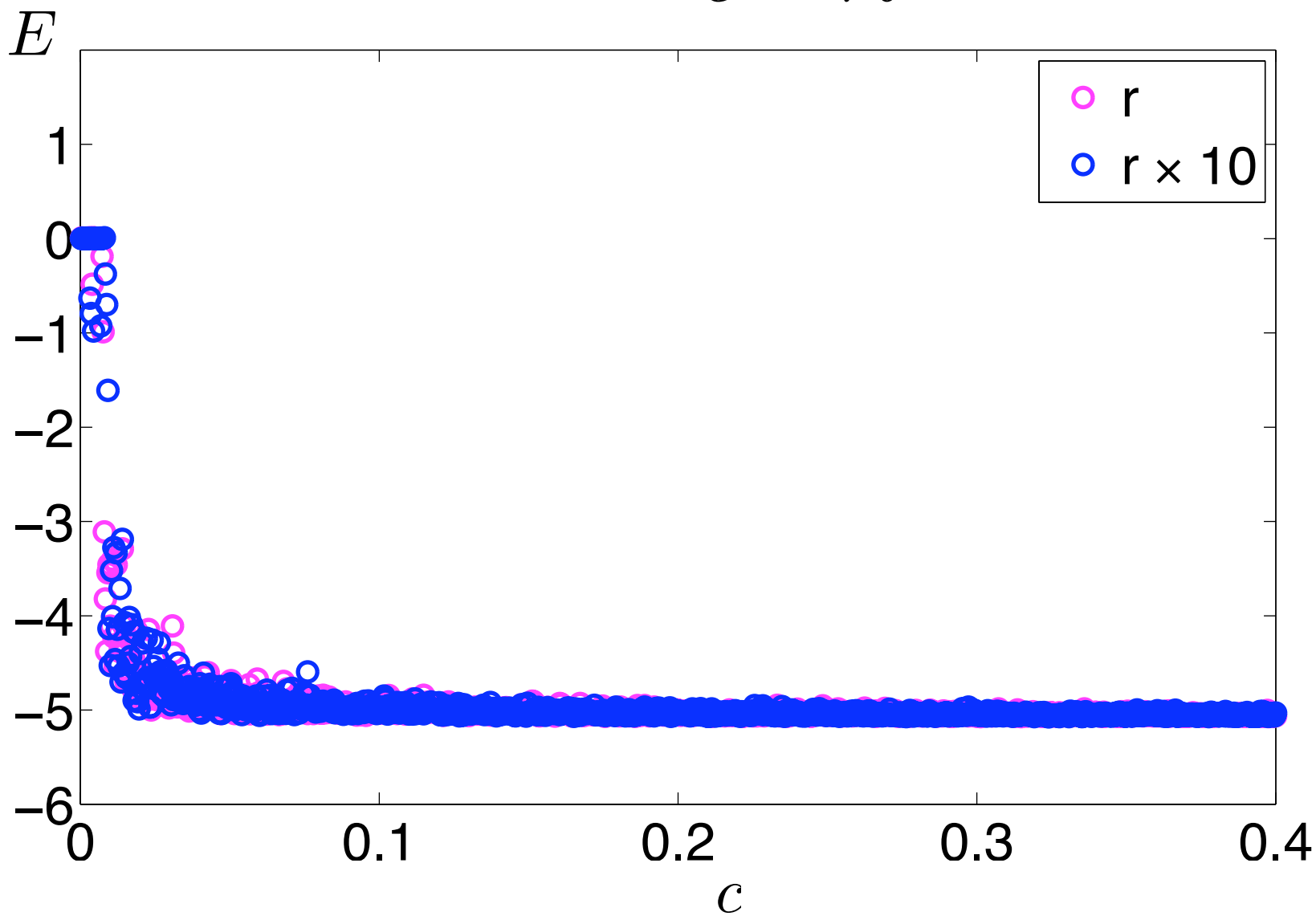
# Rate Dependence

$$c = rt$$



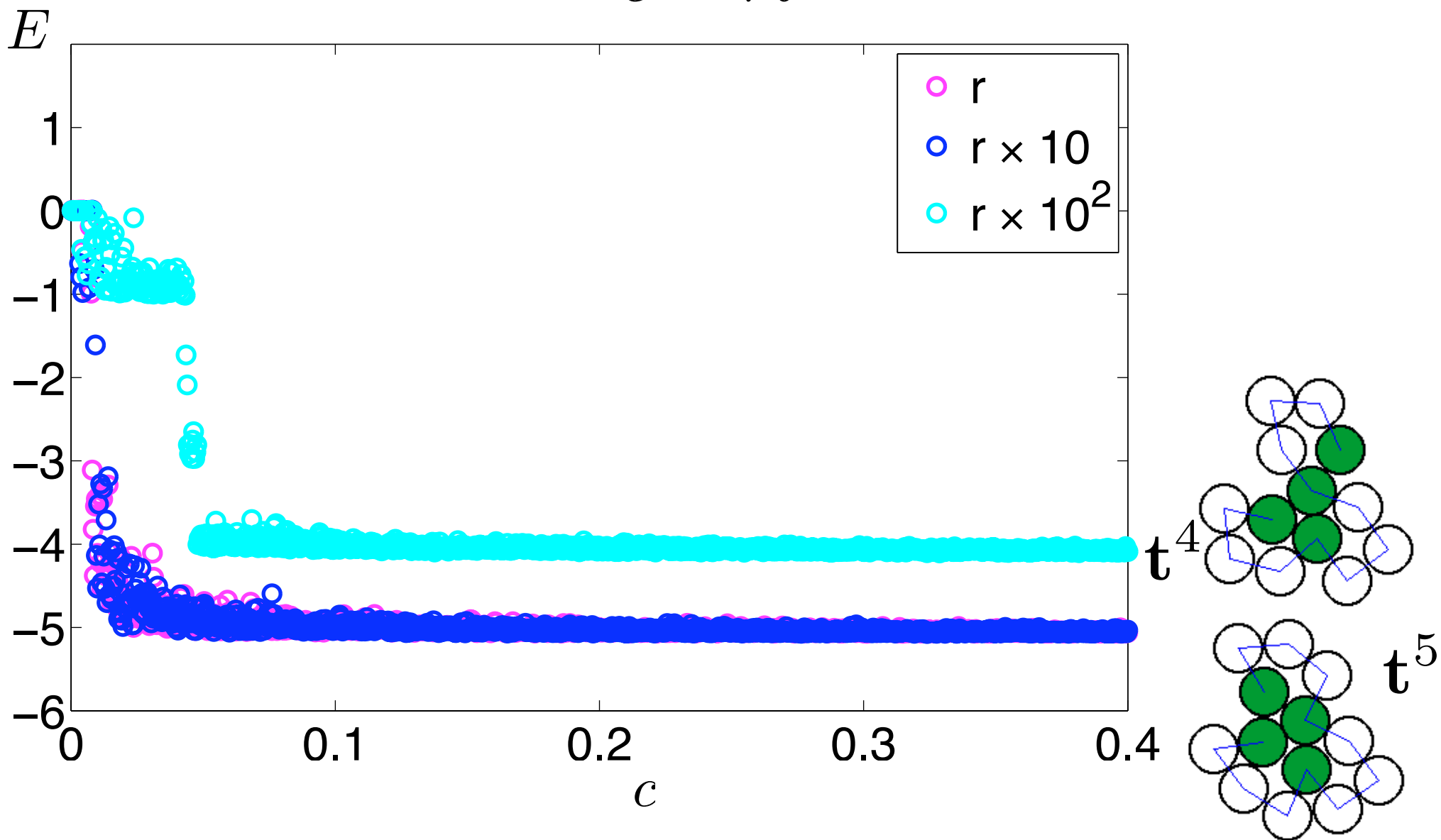
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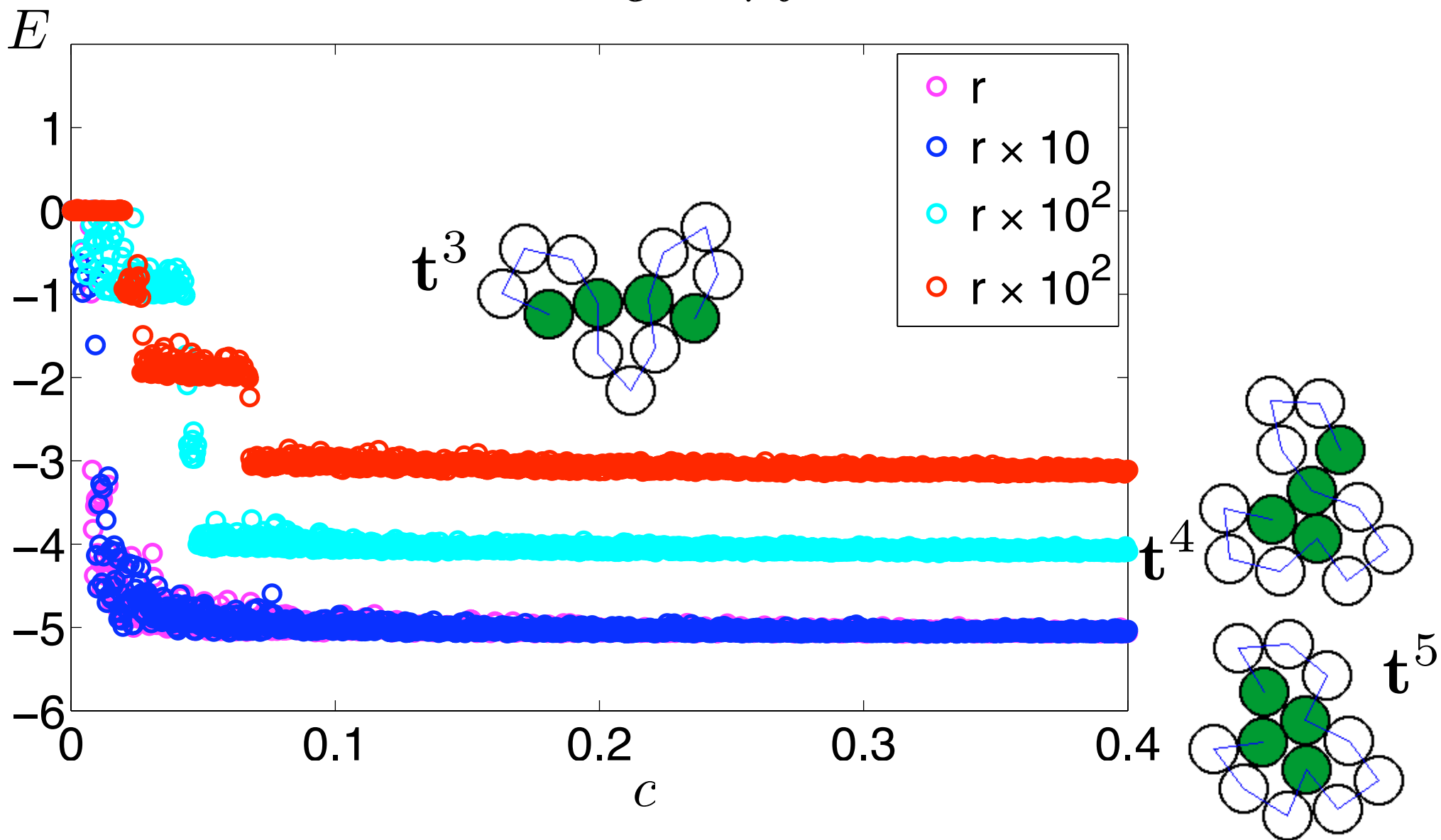
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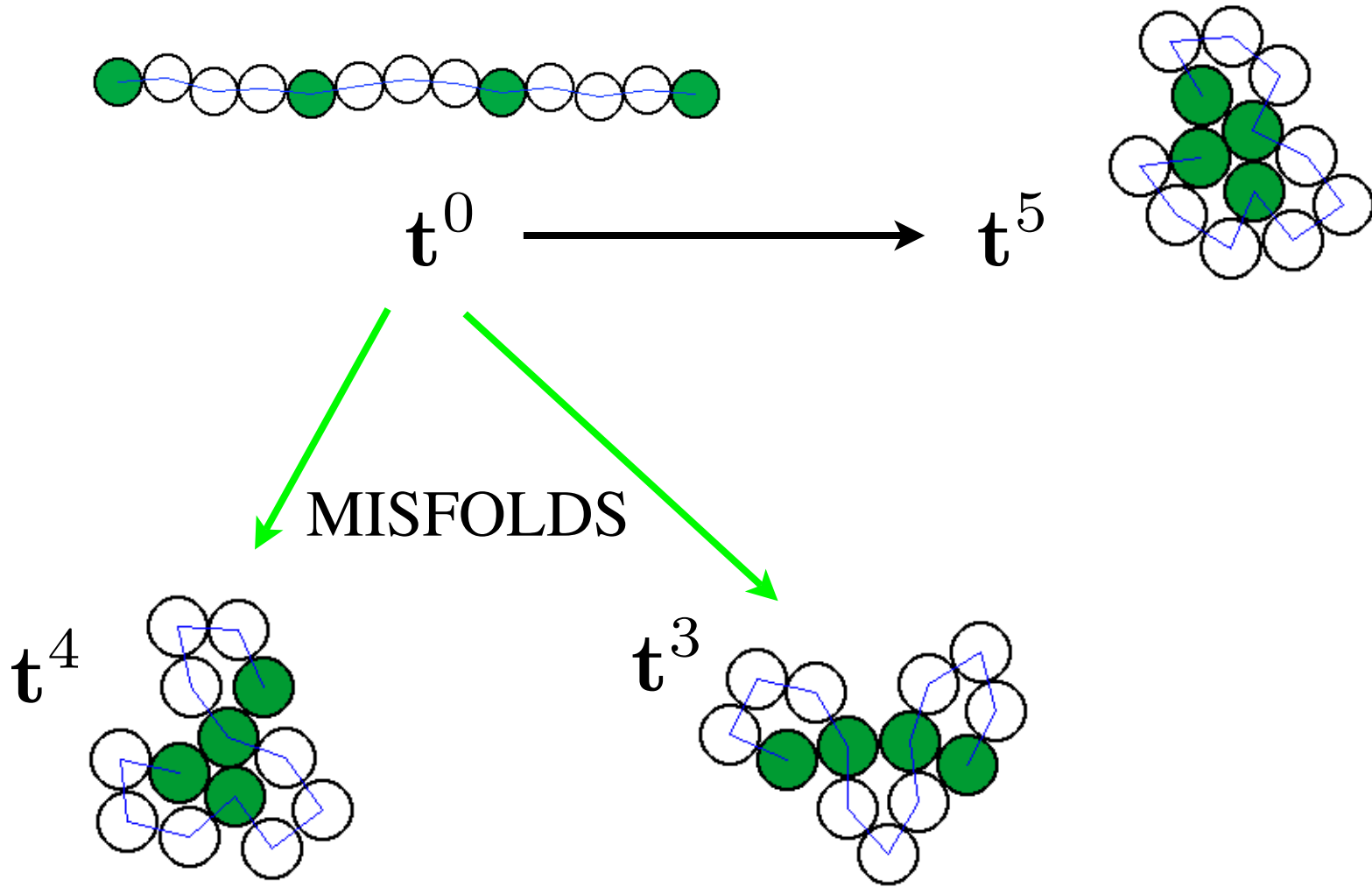


# Rate Dependence

$$c = rt$$



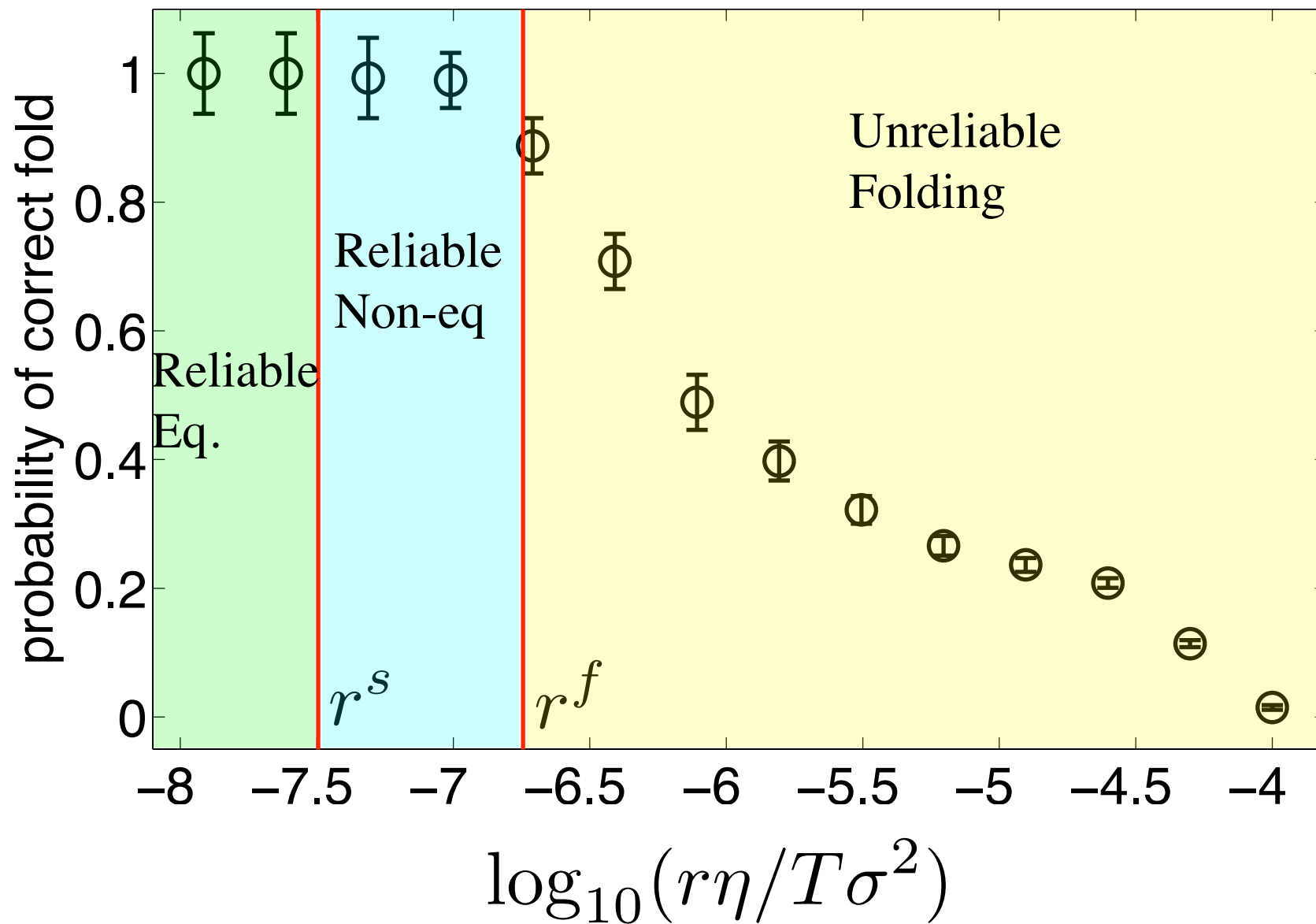
# Free Energy Reaction Path



We calculate  $r^f$  and  $r^s$  for this transition



# Reliable Folding



# Summary

- Reliable Folding can occur on Non-Funneled Energy Landscapes, depending on the folding-induction rate  $r$ .
- The Free Energy Reaction Path is a useful framework to understand the dynamics.
- Folding can occur either as an equilibrium process or out of equilibrium.
- The native state can be a global or local minimum of free energy.
- These predictions can be tested in simulations and experiments by investigating rate dependence.